Chemistry 2 Production of Materials - Part 2 Study online at guizlet.com/_lvyw3c Quizlet

1. electrode	a conductor, usually a metal or graphite, through which an electrical current can enter or leave a medium, such as a solution	20. isomer	a compound that contains the same numbers and kinds of atoms as another, but has a different structure
2. electrolyte	an aqueous solution or molten substance that contains ions that are free to move and allows charges to move between electrodes	21. isotopes	atoms of the same element that have the same atomic number but different mass number i.e. they have the same number of protons, but a different number of neutrons
3. electromotive force (emf)	the potential voltage difference difference across the electrodes of a galvanic cell when a	 22. IUPAC nomenclature 23. mass number 24. molar heat of combustion 	
	negligibly small current is being drawn		the system provided by the IUPAC for clearly naming chemicals with an explicit or implied relationship to the structure of
5 enzymes	a reaction that proceeds by absorbing near		
	in all organisms		compounds
6. feedstock	a reactant used in an industrial process that is made from raw materials		the nucleus of an atom
7. fermentation	the decomposition of carbohydrates by micro-organisms in the absence of air		mole of a substance is completely combusted
8. fermenting	the breaking down of complex molecules brought about by some state of activity, such as when grape sugar is changed into ethanol	25. monomer	the small chained chemical unit that must be linked over and over to form a larger chain
9. fissionable	a nucleus that is able to be split	26. monosaccharides	carbohydrates that cannot be hydrolysed into two or more simpler sugars, and are known as single sugar units
10. fossil fuel	a naturally occurring energy source formed within the Earth's crust from decomposing		
11 function	plant and animal matter		
The Traction	a chemical compound separated from a large number of compounds within a mixture		
12. fractional distillation	a process by which substances are vaporised and then condensed by cooling		
13. free radicals	atoms or molecules that are reactive as they have unpaired electrons		
14. fuel cell	an electric cell that converts the chemical energy of a fuel, such as hydrogen, directly into electrical energy without the need to burn the fuel		
^{15.} functional group	a number of connected atoms whose presence in a molecule gives rise to characteristic chemical properties		
16. galvanic cell	an electrochemical cell in which the spontaneous occurrence of electrode reactions produces electrical energy		
17. glucose	(C6H12O6) a naturally occurring six-carbon sugar		
18. homologous series	a family of similar carbon compounds with differing number of CH2 groups and containing the same functioning group		
19. hydrocarbon	a carbon compound, also known as an organic compound, which is composed of only carbon and hydrogen atoms		